

Strong Correlations and Spin-Orbit Coupling

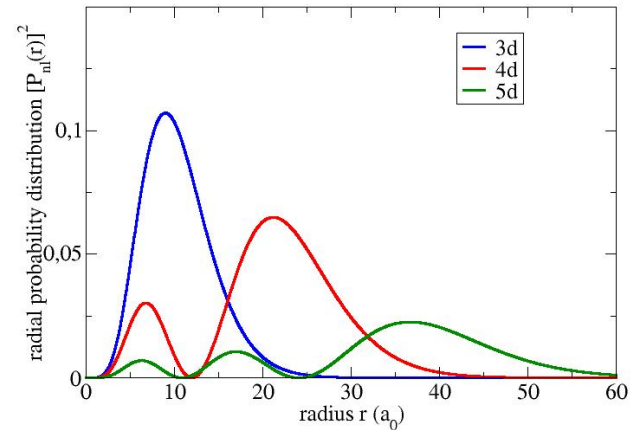
From Models to Iridates and Osmates

Markus Aichhorn

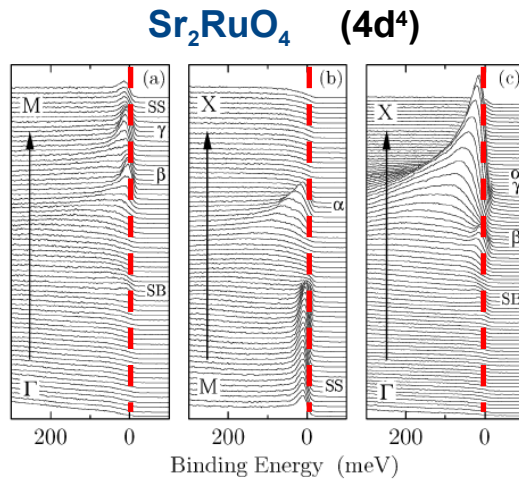
Institute of Theoretical and Computational Physics

3d	21 Sc [Ar]4s ² 3d ¹ scandium 44.96	22 Ti [Ar]4s ² 3d ² titanium 47.88	23 V [Ar]4s ¹ 3d ⁴ vanadium 50.94	24 Cr [Ar]4s ¹ 3d ⁵ chromium 52.00	25 Mn [Ar]4s ² 3d ⁵ manganese 54.94	26 Fe [Ar]4s ² 3d ⁶ iron 55.85	27 Co [Ar]4s ¹ 3d ⁷ cobalt 58.93	28 Ni [Ar]4s ² 3d ⁸ nickel 58.69	29 Cu [Ar]4s ¹ 3d ¹⁰ copper 63.55	30 Zn [Ar]4s ² 3d ¹⁰ zinc 65.39
4d	39 Y [Kr]5s ² 4d ¹ yttrium 88.91	40 Zr [Kr]5s ² 4d ² zirconium 91.22	41 Nb [Kr]5s ¹ 4d ⁴ niobium 92.91	42 Mo [Kr]5s ¹ 4d ⁵ molybdenum 95.94	43 Tc [Kr]5s ² 4d ⁵ technetium (98)	44 Ru [Kr]5s ¹ 4d ⁶ ruthenium 101.1	45 Rh [Kr]5s ¹ 4d ⁷ rhodium 102.9	46 Pd [Kr]4d ¹⁰ palladium 106.4	47 Ag [Kr]5s ¹ 4d ¹⁰ silver 107.9	48 Cd [Kr]5s ² 4d ¹⁰ cadmium 112.4
5d	57 La* [Xe]6s ² 5d ¹ lanthanum 138.9	72 Hf [Xe]6s ² 4f ¹⁴ 5d ² hafnium 178.5	73 Ta [Xe]6s ² 4f ¹⁴ 5d ³ tantalum 180.9	74 W [Xe]6s ² 4f ¹⁴ 5d ⁴ tungsten 183.9	75 Re [Xe]6s ² 4f ¹⁴ 5d ⁵ rhenium 186.2	76 Os [Xe]6s ² 4f ¹⁴ 5d ⁶ osmium 190.2	77 Ir [Xe]6s ² 4f ¹⁴ 5d ⁷ iridium 190.2	78 Pt [Xe]6s ¹ 4f ¹⁴ 5d ⁹ platinum 195.1	79 Au [Xe]6s ¹ 4f ¹⁴ 5d ¹⁰ gold 197.0	80 Hg [Xe]6s ² 4f ¹⁴ 5d ¹⁰ mercury 200.5

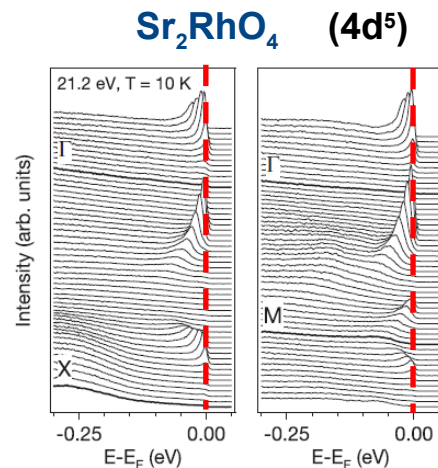
4d states are **less** localized/correlated **than** 3d
5d states are **less** localized/correlated **than** 4d



Angle Resolved Photo-Emission Spectroscopy (ARPES):

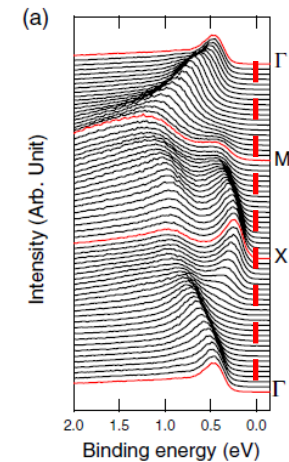


Damascelli *et al*,
Phys Rev Lett 85, 5194 (2000)



Baumberger *et al*,
Phys Rev Lett 96, 246402 (2006)

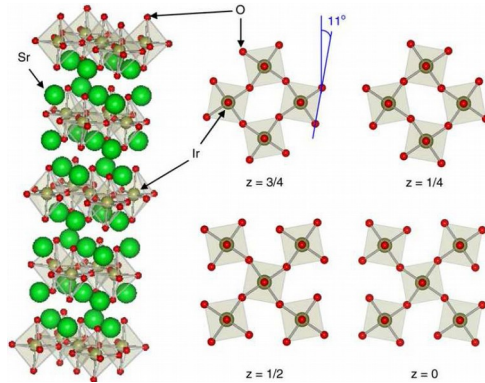
Sr₂IrO₄ (5d⁵)



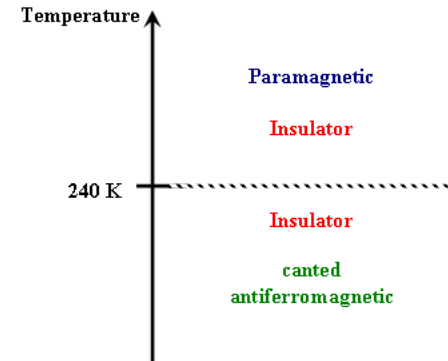
Kim *et al*,
Phys Rev Lett 101, 076402 (2009)

Sr_2IrO_4 : Structure similar to La_2CuO_4 but with distortions

IrO_6 octahedra are rotated around the z axis by about 11°



Klein & Terasaki, J. Phys.: Cond. Mat.20 (2008)



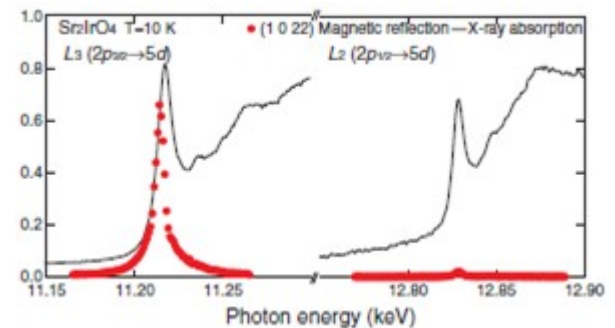
Moon *et al*, Phys Rev B 80, 195110 (2009)

Insulator at all temperatures, but open Ir d^5 shell.

Optical gap at 300 K (room temperature) of 0.26 eV.

Important role of the spin-orbit coupling

$$\zeta_{\text{SO}} \sim 0.4 \text{ eV}$$

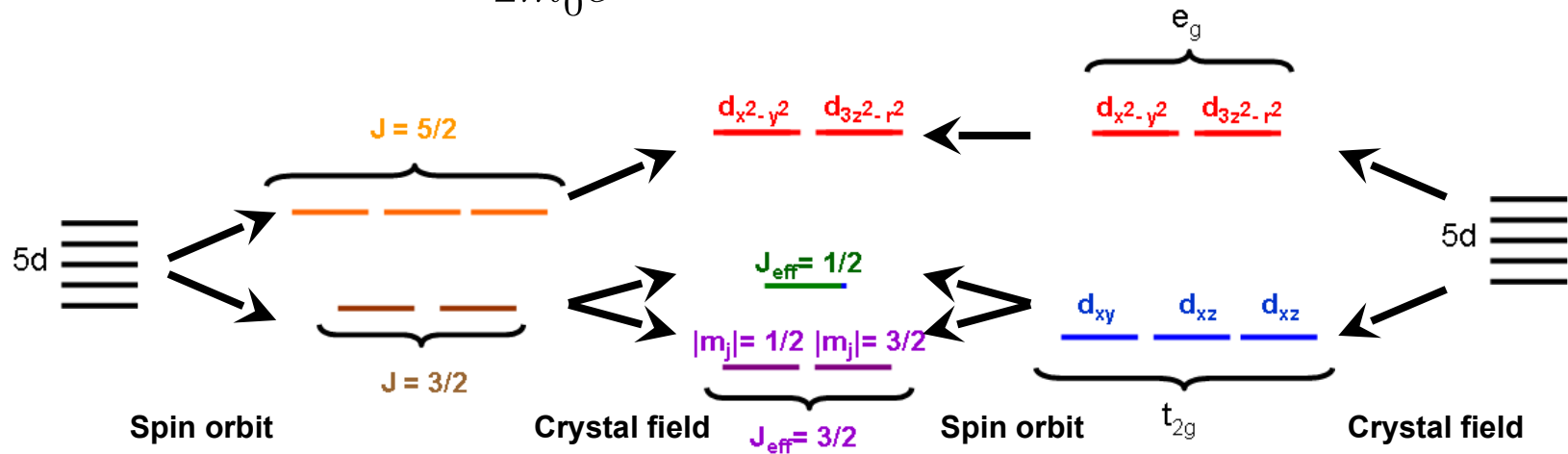


Kim *et al*, Science 323, 1329 (2009)

Spin-Orbit Coupling and local Hamiltonian

Relativistic correction of Schrödinger equation

$$H_{SO} = \frac{\hbar^2}{2m_0^2c^2} \mathbf{S} \cdot [\nabla V(\mathbf{r}) \times \mathbf{p}] = \zeta_{SO} (\mathbf{L} \cdot \mathbf{S})$$



Crystal field induced by the oxygen : 5d states are split into e_g , $J_{\text{eff}} = 1/2$ and $J_{\text{eff}} = 3/2$ multiplets.

A model approach: DMFT on a Bethe lattice

Kanamori 3-band Hamiltonian

$$H_I = \sum_i U n_{i\uparrow} n_{i\downarrow} + (U - 2J) \sum_{i \neq j} n_{i\uparrow} n_{j\downarrow} + (U - 3J) \sum_{i < j, \sigma} n_{i\sigma} n_{j\sigma} \\ + J \sum_{i \neq j} c_{i\uparrow}^\dagger c_{j\downarrow}^\dagger c_{i\downarrow} c_{j\uparrow} + J \sum_{i \neq i} c_{i\uparrow}^\dagger c_{i\downarrow}^\dagger c_{j\downarrow} c_{j\uparrow}$$

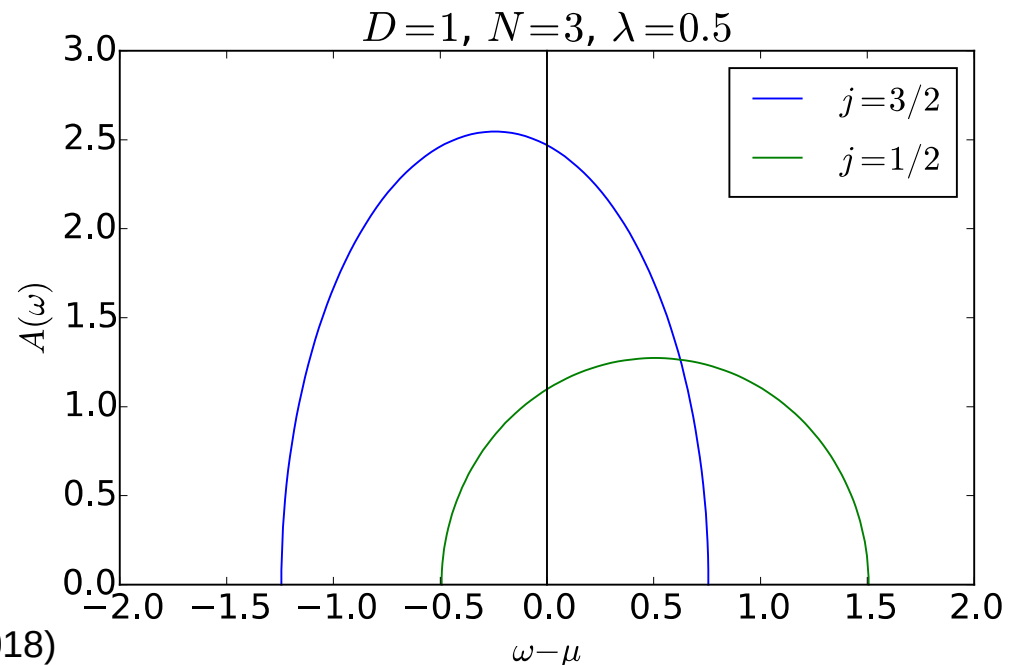
Bethe lattice

SOC Hamiltonian

$$H_{SO} = \lambda \mathbf{L}_{t2g} \cdot \mathbf{S}$$

Hamiltonian diagonal
in j -basis

Weak sign problem in CTHYB



R. Triebl, ... M. Aichhorn, PRB 98, 205128 (2018)

Atomic limit: the j -basis

Diagonalization leads to j -basis (only $j = 1/2$ terms):

$$|j = \frac{1}{2}, m_j = -\frac{1}{2}\rangle = \frac{1}{\sqrt{3}}|d_{xy}^{\downarrow}\rangle + \frac{i}{\sqrt{3}}|d_{xz}^{\uparrow}\rangle - \frac{1}{\sqrt{3}}|d_{yz}^{\uparrow}\rangle$$

$$|j = \frac{1}{2}, m_j = \frac{1}{2}\rangle = \frac{1}{\sqrt{3}}|d_{xy}^{\uparrow}\rangle + \frac{i}{\sqrt{3}}|d_{xz}^{\downarrow}\rangle + \frac{1}{\sqrt{3}}|d_{yz}^{\downarrow}\rangle$$

Advantage: Single-particle part of hamiltonian is diagonal

Disadvantage: Interaction part becomes involved

Pre-factors of interaction terms:

	one orbital	opp. spin	aligned spin
Cubic, crystal field	U	$U - 2J$	$U - 3J$
j -basis	$U - 4/3 J$	$U - 7/3 J$	$U - 7/3 J$

Atomic limit and the Crystal-Field Analogue

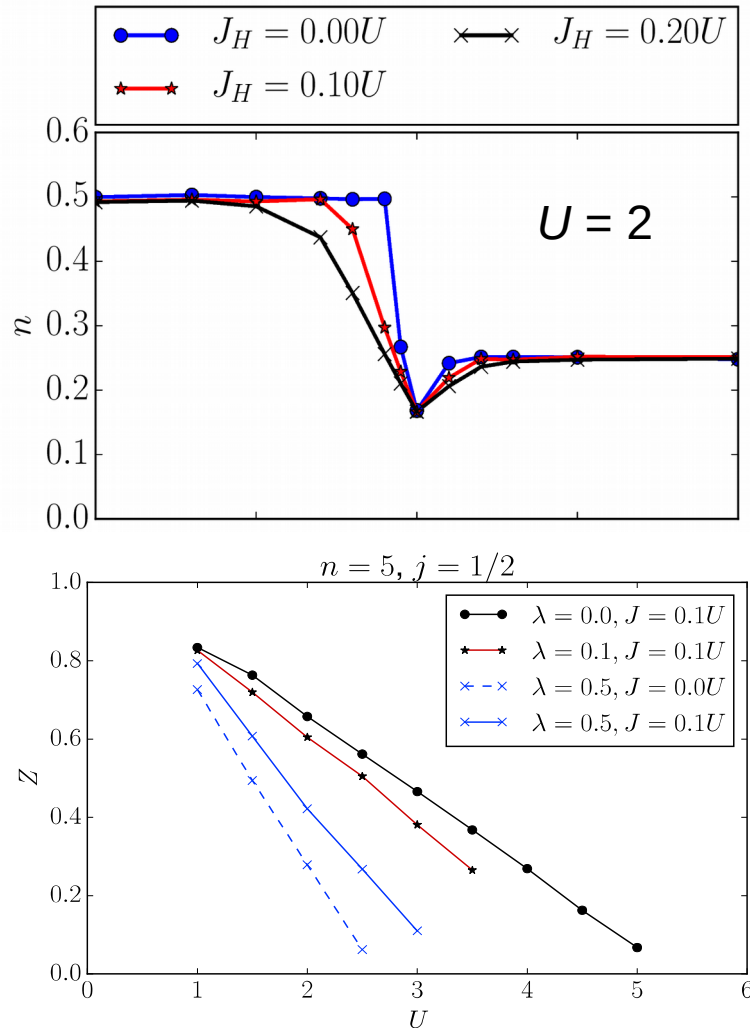
Limit of large SOC / Crystal field:

N	SOC	Crystal field
1	$U - 7/3 J$	$U - 3 J$
2	$U - J$	$U + J$
3	$U - 7/3 J$	$U - 3 J$
4	$U - 3 J + 3/2 \lambda$	$U - 5 J + 3/2 \Delta$
5	$U - 4/3 J$	U

Qualitative differences for $N = 2$ and $N = 5$!

To what extent is this difference reflected in a true calculation?

The easy cases: 1 electron or 1 hole



Main effect:

Reduction of number of orbitals

Large SOC limit:

$N = 1$: One electron in two bands

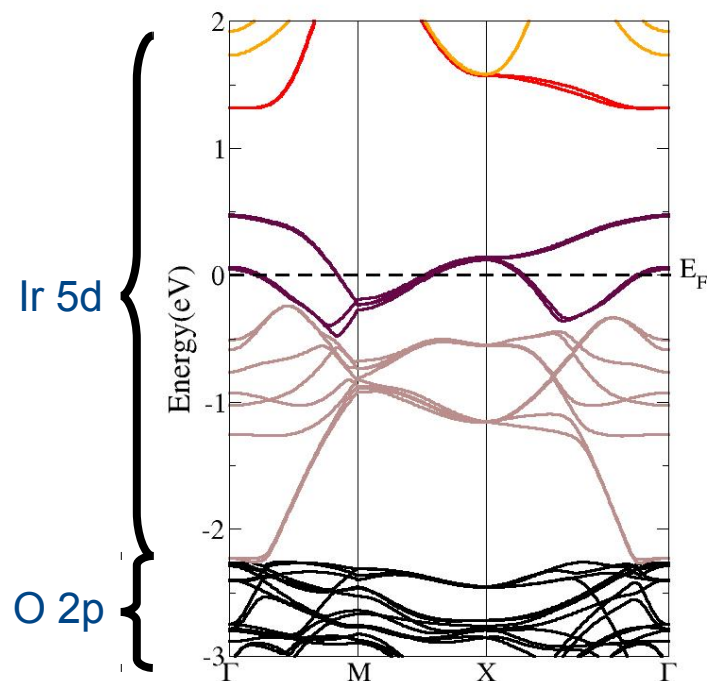
$N = 5$: One electron in one band

Hunds coupling prevents polarization

Sizable Effect for moderate SOC strength!

Depends on J due to transformation of hamiltonian!

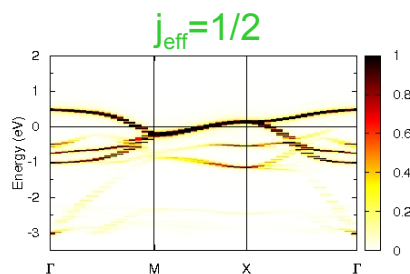
1 hole: Iridate Sr_2IrO_4



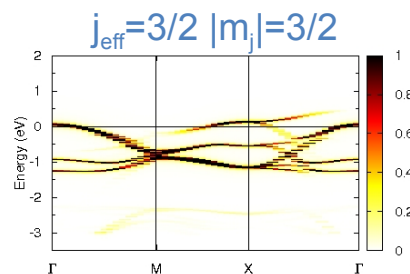
4 bands at Fermi level:
metallic Kohn-Sham band structure

$$j_{\text{eff}} = 1/2$$

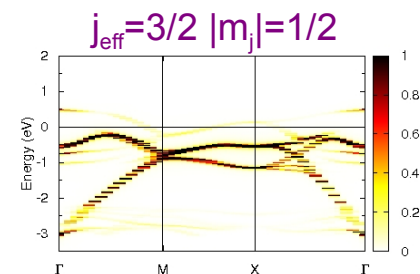
$$j_{\text{eff}} = 3/2$$



$$n = 1.14$$



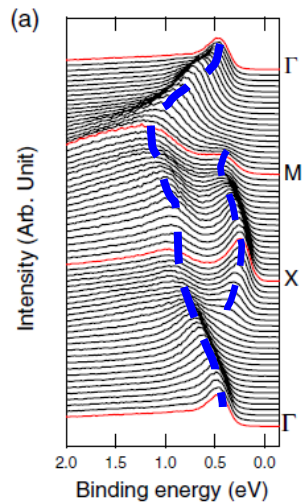
$$n = 1.96$$



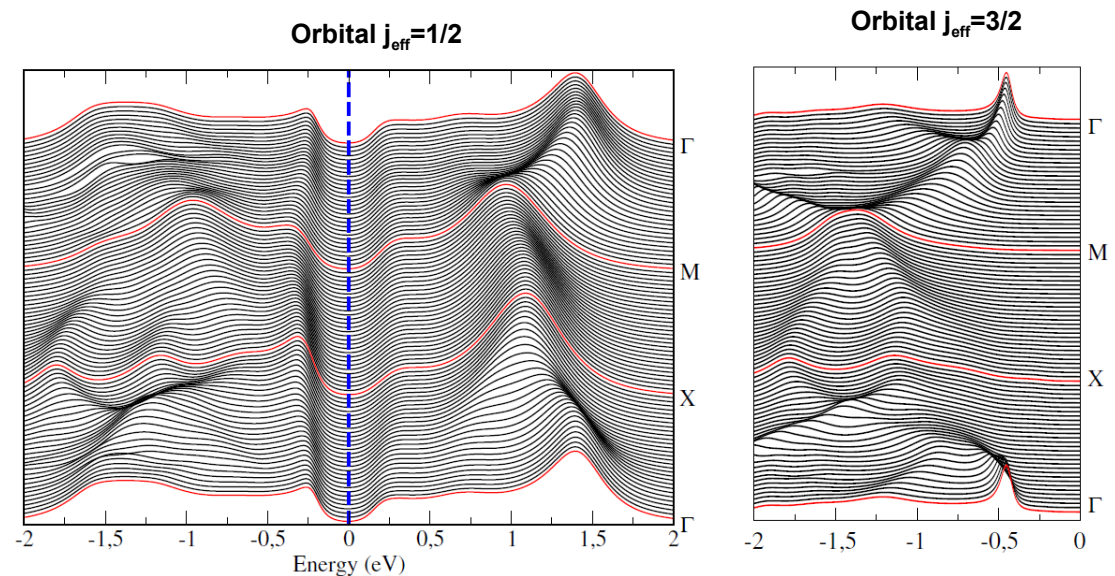
$$n = 2.00$$

Experimental spectrum

Kim *et al*, PRL 101, 076402 (2009)



DMFT, orbitally resolved spectral density



C. Martins, M. Aichhorn, et al., PRL 107, 266404 (2012)

Sr_2IrO_4 is a $j_{\text{eff}}=1/2$ Mott insulator ($U = 2.2$ eV and $J = 0.3$ eV)

neither a magnetic order, nor an orbital order but a “spin-orbital order”

The suppression of spin-orbital fluctuations due to:

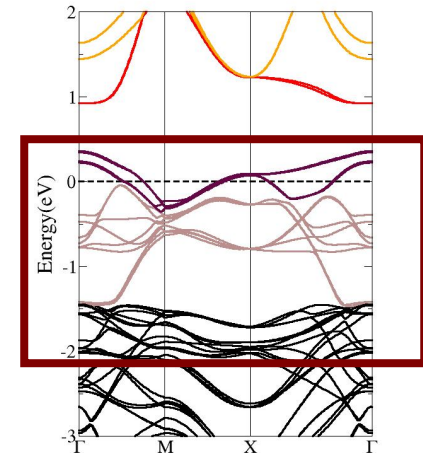
- Spin orbit coupling ($\zeta_{\text{SO}} \sim 0.4$ eV)
- the structural distortions (rotation of 11° of the IrO_6 octahedra)

Iridate vs. Rhodate: Effect of spin-orbital polarization

Sr_2RhO_4 has identical crystal structure as Sr_2IrO_4

But smaller SO coupling 0.2

Charge repartition between the $j_{\text{eff}}=1/2$ state and the two $j_{\text{eff}}=3/2$ states are different.



Charge	Sr_2IrO_4	Sr_2RhO_4
$j_{\text{eff}} = 1/2$	1.14	1.40
$j_{\text{eff}} = 3/2 \ m_j = 1/2$	2.00	1.96
$j_{\text{eff}} = 3/2 \ m_j = 3/2$	1.96	1.64

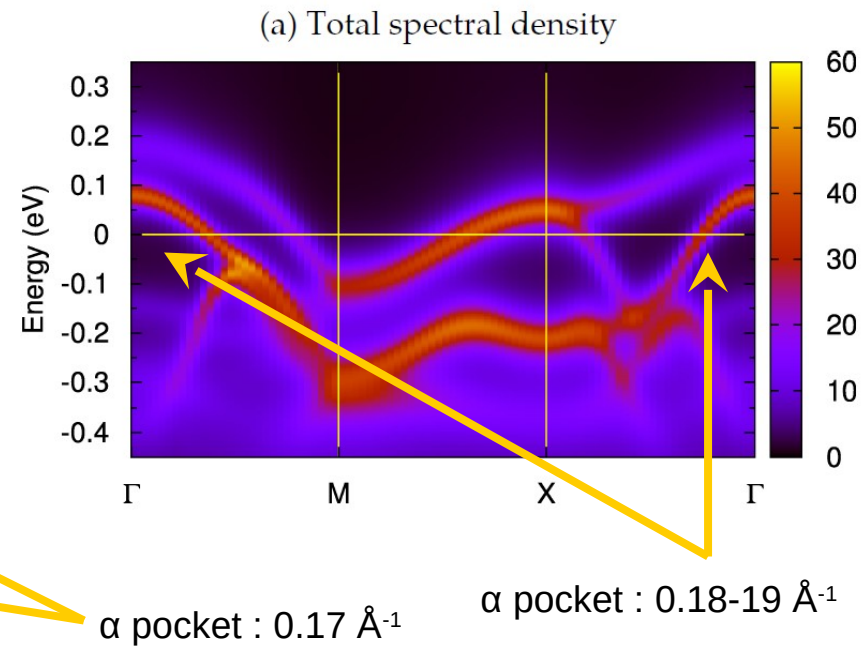
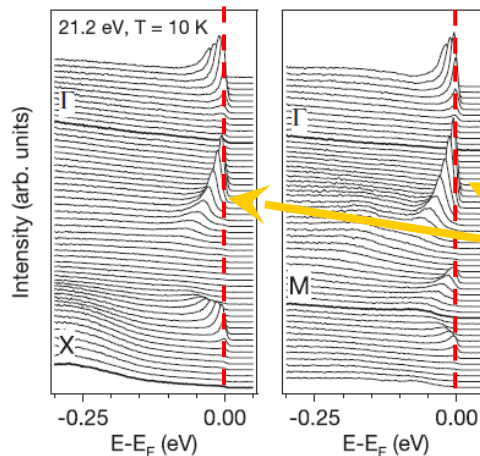
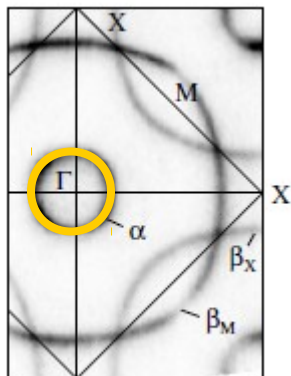
Sr_2RhO_4 is a “three-quarter-filled two-bands” system

Sr_2RhO_4 is a METAL!

Iridate vs. Rhodate: Effect of spin-orbital polarization

Sr_2RhO_4 is a
paramagnetic correlated metal
 ($U = 1.6$ eV and $J = 0.3$ eV)

Comparison with experimental spectral function
 obtained with ARPES (at 10K) :



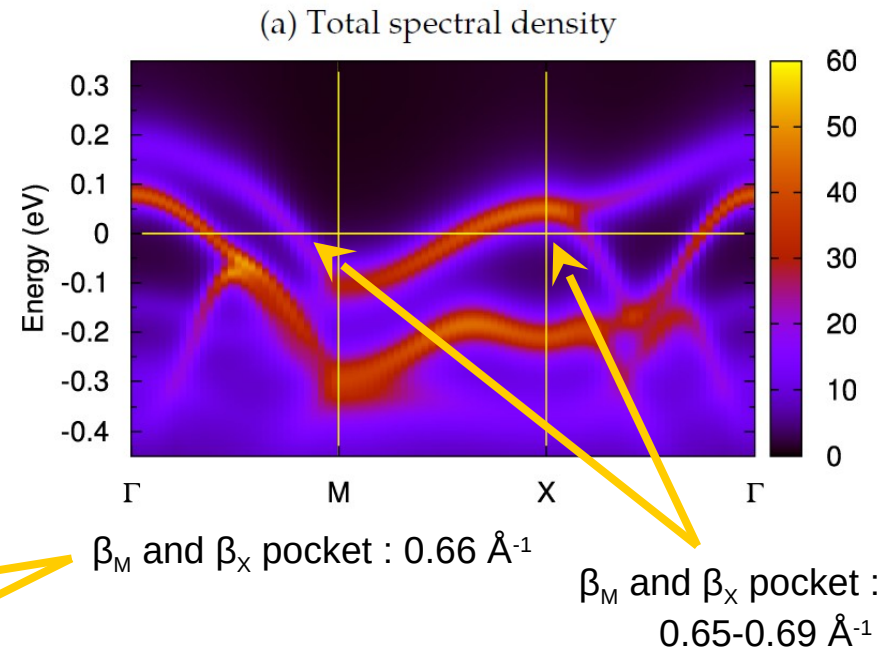
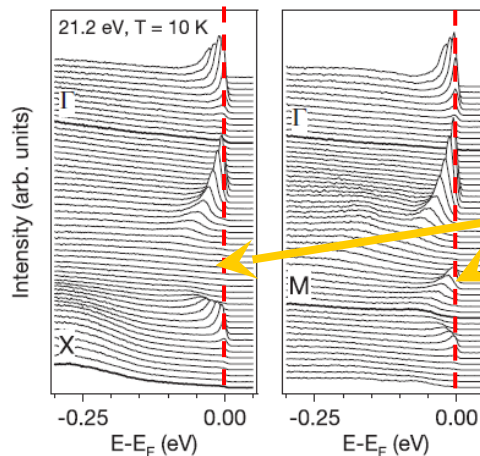
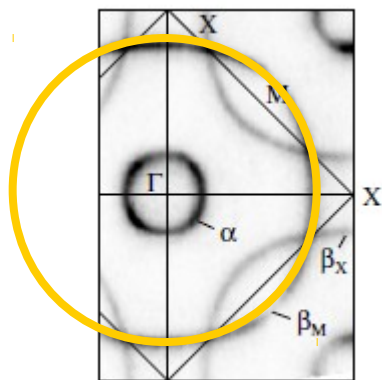
Baumberger et al.
 Phys Rev Lett 96, 246402 (2006)

C. Martins, M. Aichhorn et al., JPCM 2017
 PsiK highlight of the month, Nov 2016

Iridate vs. Rhodate: Effect of spin-orbital polarization

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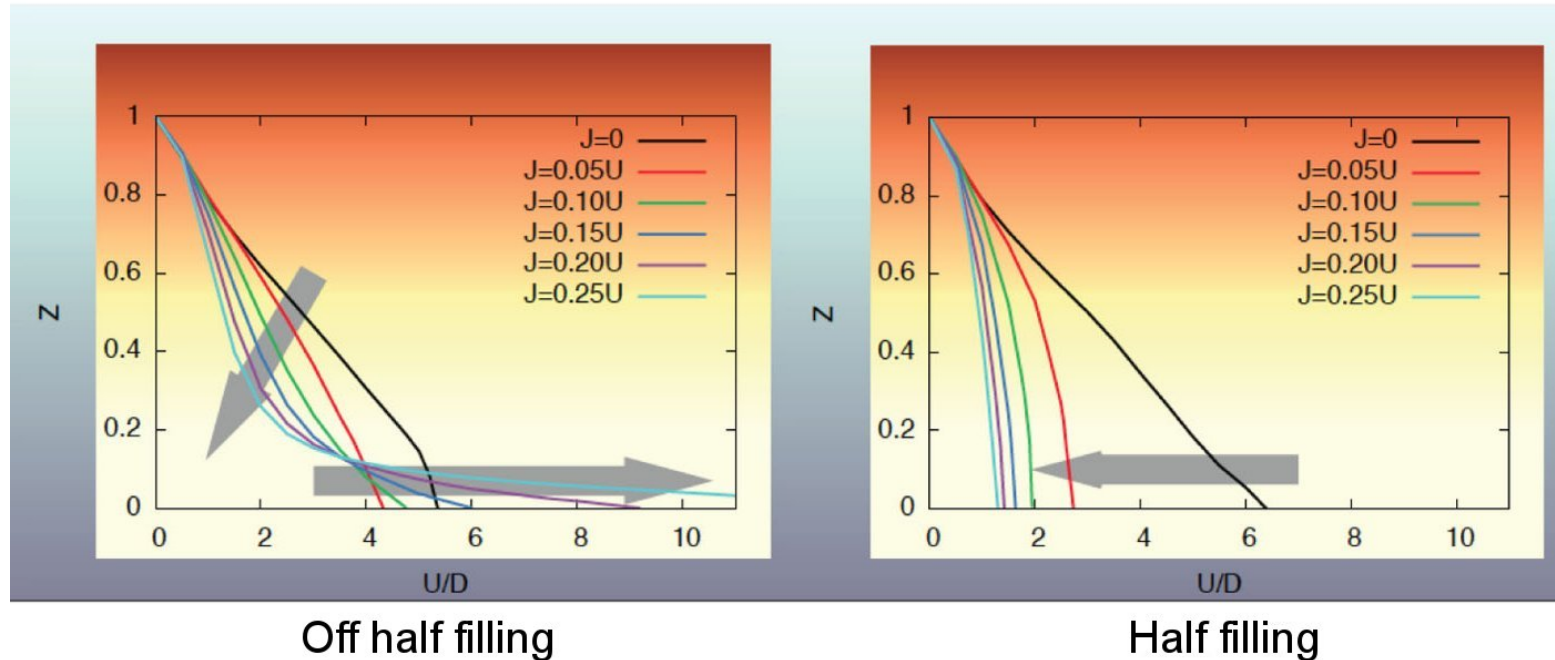


Baumberger et al.
 Phys Rev Lett 96, 246402 (2006)

C. Martins, M. Aichhorn et al., JPCM 2017
 PsiK highlight of the month, Nov 2016

2 electrons in 3 bands: Hund's metals

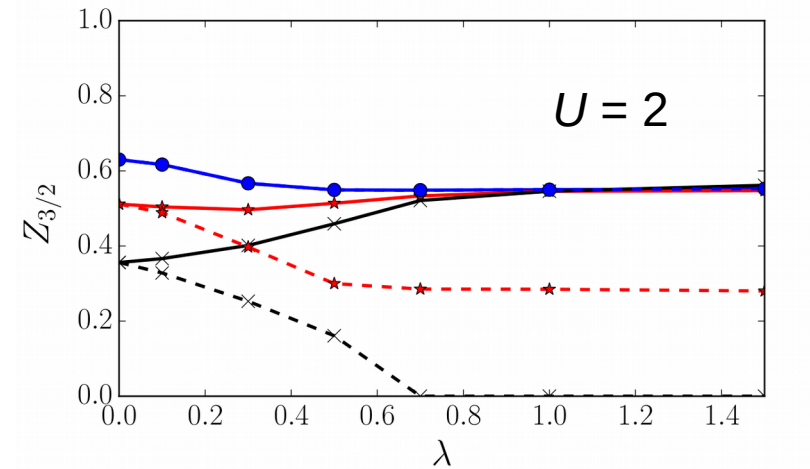
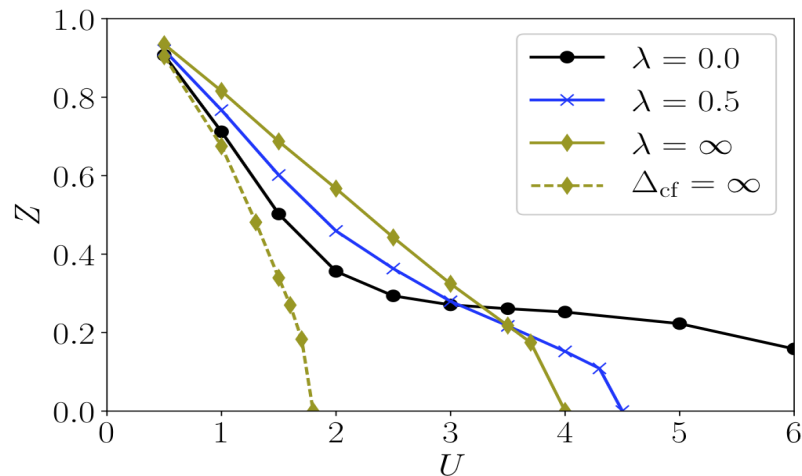
L. de Medici *et al.*, PRL 107, 256401 (2011)



Two antagonistic effects: Lowering coherence scale
 Strong influence on metal/insulator transition

How does this change with spin orbit?

2 electrons in 3 bands: Hund's metals

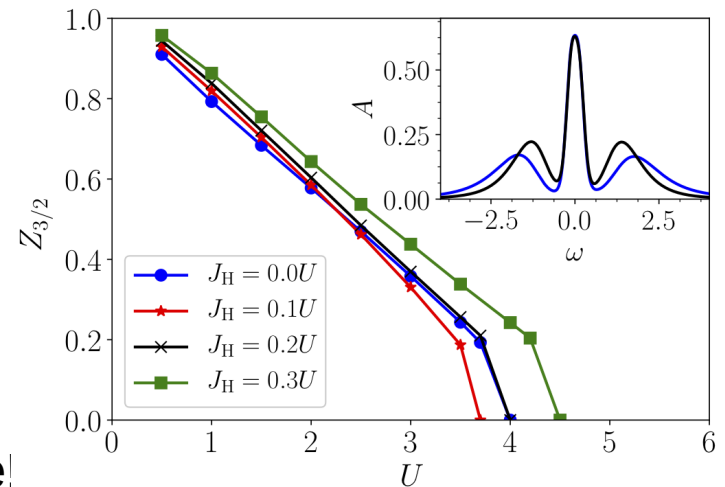


Hunds metal "tail" disappears

System becomes half filled

BUT: Large SOC: weak dependence on J !
Crystal field very different!

Atomic Hamiltonian:
after transformation very different Hund's rule!



2 electrons in 3 bands: Hund's metals

Without SOC:

Insulator at $N=3$

$N=2$ as doped Mott insulator

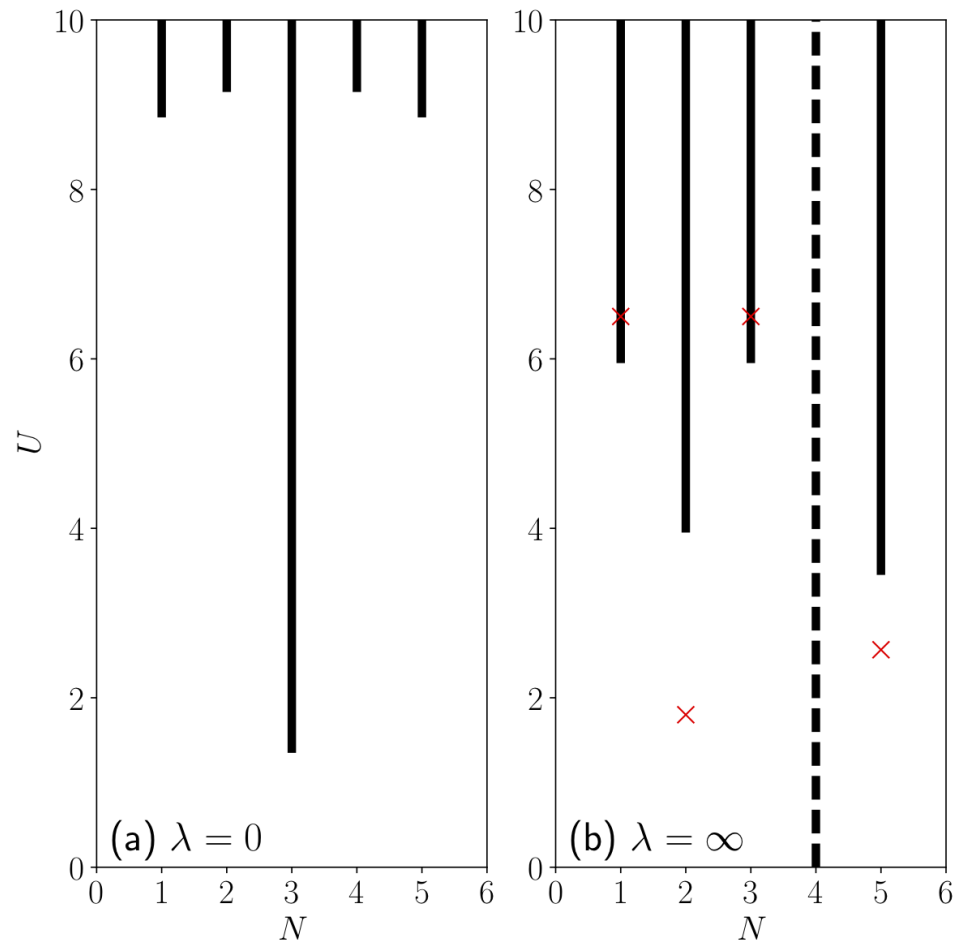
Hunds tail

With SOC:

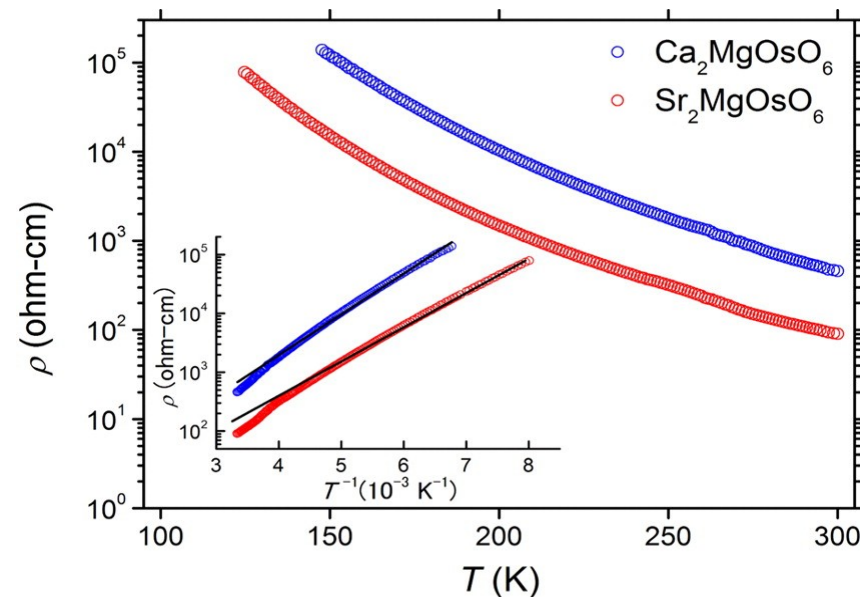
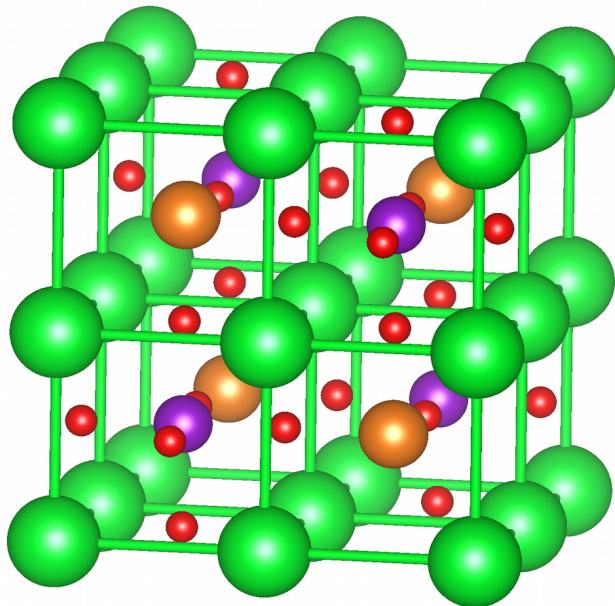
Insulator at $N=2$

No Mott state close to metal

No Hunds tail



2 electrons in 3 bands: Osmates $\text{Sr}_2\text{MgOsO}_6$



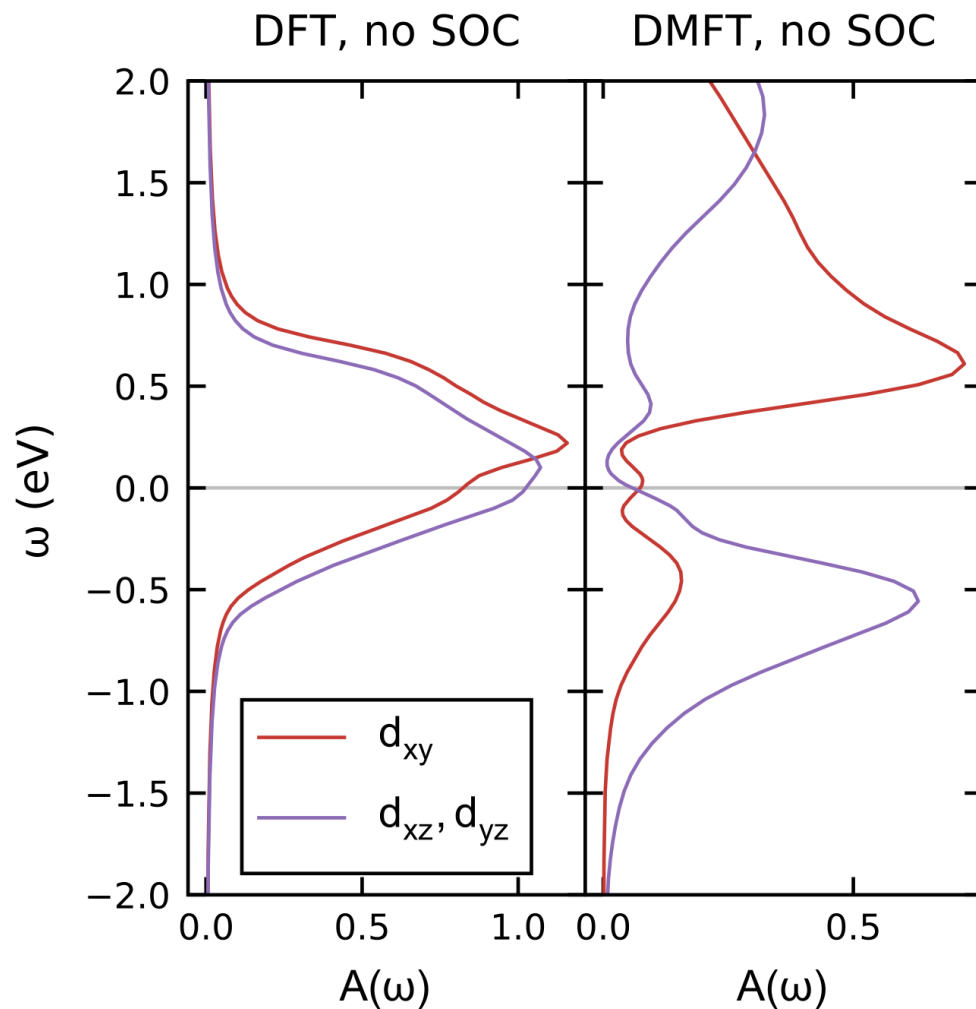
Y. Yuan et al., Inorg Chem. 54, 3422 (2015)

Double perovskite

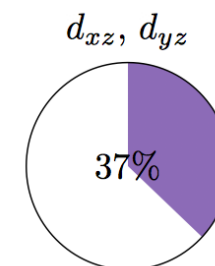
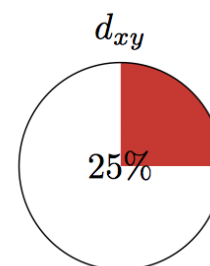
Slightly distorted cages

“These double-perovskite oxide are likely to be Mott insulators”

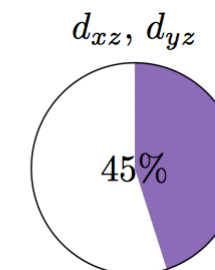
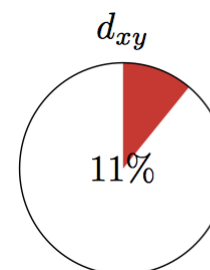
Osmate $\text{Sr}_2\text{MgOsO}_6$ without SOC



DFT:

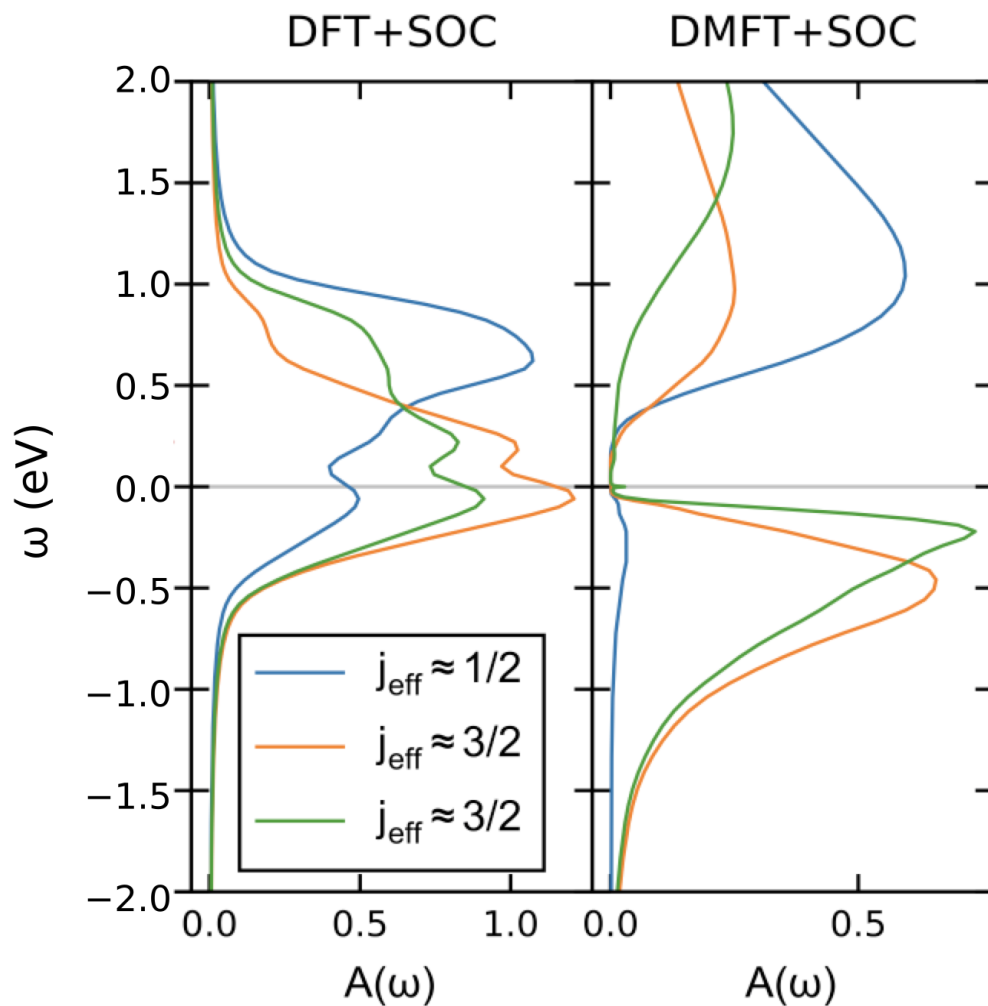


DMFT: bad metal

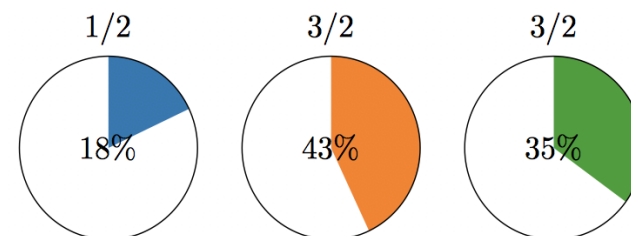


G. Kraberger and M. Aichhorn, in prep.

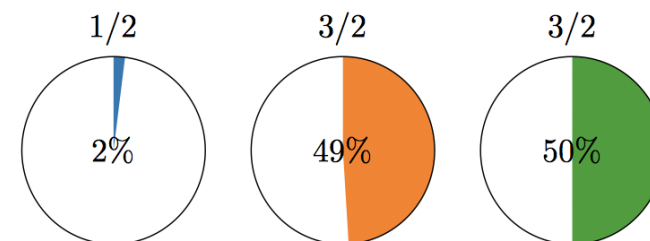
Osmate $\text{Sr}_2\text{MgOsO}_6$ with SOC



DFT: j-basis



DMFT: polarised insulator



The self energy and effective SOC

$$\Sigma_a = \frac{2}{3}\Sigma_{\frac{3}{2}} + \frac{1}{3}\Sigma_{\frac{1}{2}}$$

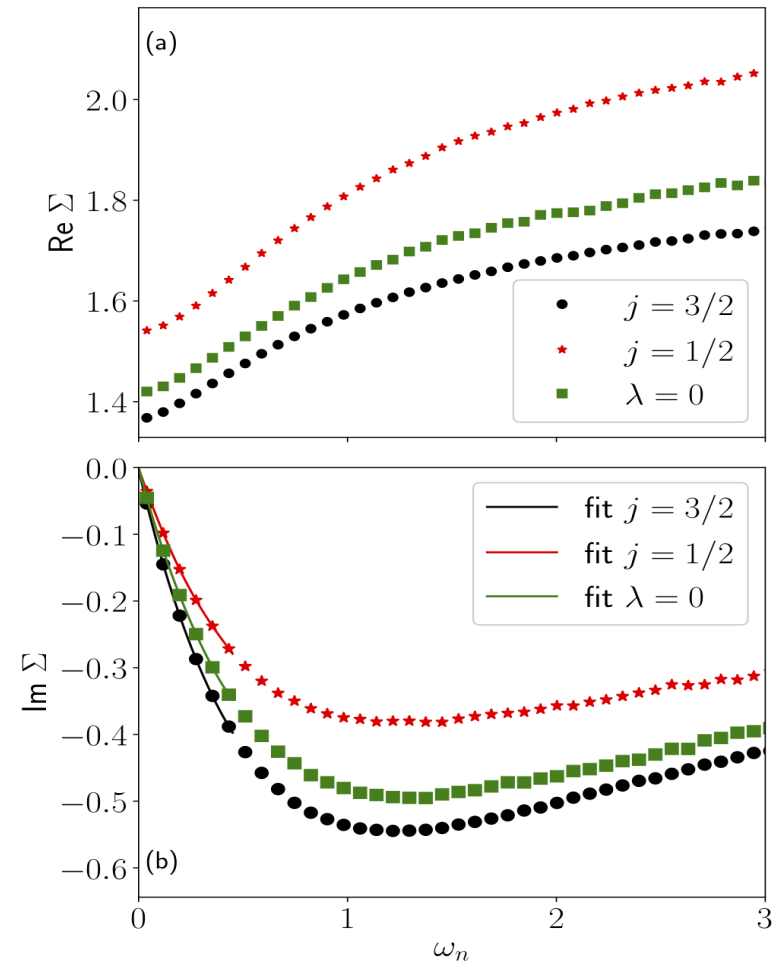
$$\Sigma_d = \Sigma_{\frac{1}{2}} - \Sigma_{\frac{3}{2}}$$

$$\rightarrow \Sigma = \Sigma_a \mathbb{I} + \frac{2}{3}\Sigma_d \mathbf{l} \cdot \mathbf{s}$$

Eff. Spin-Orbit Coupling:

$$\lambda_{\text{eff}} = \lambda + \frac{2}{3} \text{Re}\Sigma_d(i\omega_n \rightarrow 0)$$

$$\lambda_{\text{eff}} > \lambda$$



Strong Correlations and Spin-Orbit Coupling

Spin-Orbit coupling important for heavy atoms

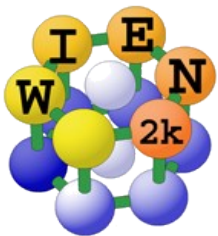
In model calculations:

- Effect strongest for $N=5$, strong reduced for $N=3$ and $N=2$
- Effective crystal field picture has to be used with care

Iridates: Spin-Orbit assisted Insulator

- Reduced Spin-Orbital Degeneracy, effective one-band model

Osmates Double Perovskites: Similar effect, but $N=2$



trigs



FWF

Strong Correlations and Spin-Orbit Coupling



Elias Assmann



Gernot Kraberger



Robert Triebel



Manuel Zingl



Cyril Martins



Loig Vaugier



Silke Biermann



Jernej Mravlje