

# 1-PAC: First application of a one-detector gamma-gamma perturbed angular correlation technique for the determination of physical-chemical properties of radioeuropium species

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A new 1-PAC method (perturbed angular correlation measured using one detector only) was developed recently for <sup>111</sup>In [1]. In this article the approach is applied to the radiolanthanide <sup>154</sup>Eu. The emission used in our measurements on the 123 keV state of <sup>154</sup>Gd was the 1274 – 123 keV cascade. The relevant part of the decay scheme is illustrated in Fig.1.

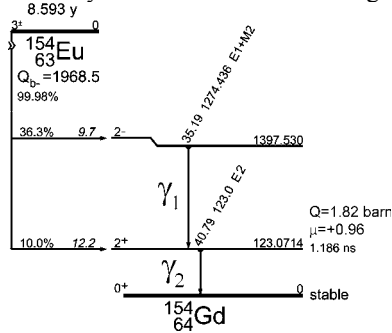


Fig.1. Relevant features of the decay scheme of <sup>154</sup>Eu

The time-integrated angular correlation of cascade (TIAC)  $\gamma$ -rays is given by  $W(\theta, \infty) = 1 + A_{22}G_2(\infty)Q_2P_2(\cos\theta) + \dots$ , [ $A_{ii}$ -angular correlation coefficients depending on spins and multiplicities of transitions;  $P_i(\cos\theta)$  - Legendre polynomials;  $Q_i$ -solid angle correction factors;  $G_i(\infty)$  -TIAC perturbation factor].

The TIAC  $W(\theta, \infty)$  of the <sup>154</sup>Eu cascade  $\gamma$ -ray coincidences are given in Fig.2. The solid and dotted lines represent the unperturbed ( $G_2(\infty) = 1$ ) and perturbed angular correlation ( $G_2(\infty) = 0.2$ ), respectively.

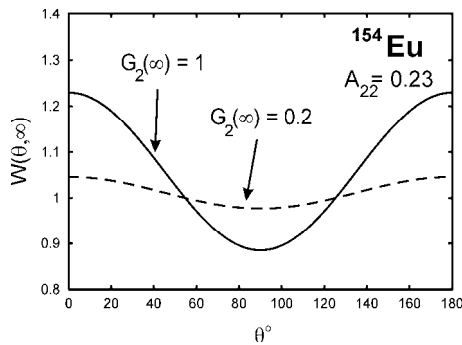


Fig.2. <sup>154</sup>Eu time-integrated angular correlation dependencies ( $A_{44} \ll A_{22}$ )

The  $\gamma$ -spectrum of <sup>154</sup>Eu (Fig. 3) consists of "mono peaks" (MP) corresponding to  $\gamma$ - and X-rays full-energy peaks and "summing peaks" (SP) corresponding to the summing energy of these radiations in different combinations. It was shown that the relative probability of recording of cascade  $\gamma$ -rays  $R_{\gamma_n+\gamma_m}$  in SP at the individual values of  $A_{ii}$ ,  $G_i(\infty)$  and  $Q_i$  is equal to TIAC  $W(\theta, \infty)$  [1]:

$$R_{\gamma_n+\gamma_m} = W_{A_{ii}, G_i, Q_i}^{\gamma_n, \gamma_m}(\theta = 0^\circ, \infty) \quad (1)$$

Similar to <sup>111</sup>In, for <sup>154</sup>Eu  $R_{\gamma_n+\gamma_m}$  can be obtained as:

$$R_{\gamma_1+\gamma_2} = N_{K_{\alpha, \beta}+\gamma_1} \cdot \frac{S_{\gamma_1+\gamma_2} \cdot S_{K_{\alpha, \beta}}}{S_{\gamma_2} \cdot S_{\gamma_1+K_{\alpha, \beta}}} \quad (2)$$

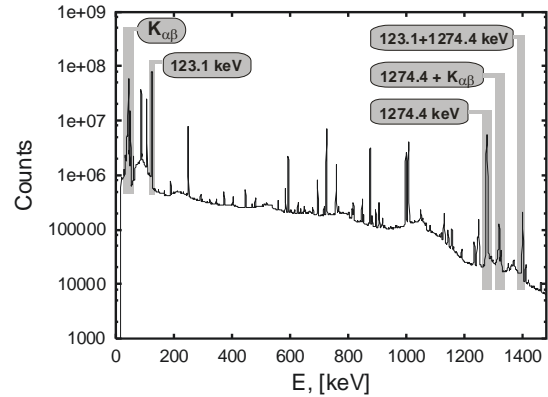


Fig.3. The  $\gamma$ -spectrum of <sup>154</sup>Eu obtained using an HPGe-detector

with  $S_{\gamma_n} - MP$  area,  $S_{\gamma_n+\gamma_m} - SP$  area. For <sup>154</sup>Eu, the normalisation coefficient is:

$$N_{K_{\alpha, \beta}+\gamma_1} = \frac{y_{e(K)}(\gamma_2)}{\sum_i y_{e(K)}(\gamma_i)} = 0.9241, \quad (3)$$

with  $y_{e(K)}(\gamma_i)$  - the yields of electrons in the conversion process on the K-shell, corresponding to  $\gamma_i$  [2].

1.20 GBq of <sup>154</sup>Eu have been produced by a 737 h neutron irradiation of 1 mg europium oxide (99.71% of <sup>153</sup>Eu) at the BER II reactor (HMI, Berlin) at a flux of  $2 \cdot 10^{14}$  n/cm<sup>2</sup> s. The ratio of <sup>155</sup>Eu/<sup>154</sup>Eu activities was obtained to be 0.39. Due to the contribution of X-rays of <sup>155</sup>Eu the overall experimental normalisation coefficient is 0.6861.

1-PAC measurements were performed using polyethylene vials ("Eppendorf") of 2 ml volume. The activity of the samples was 50 kBq. The volume of each sample was 0.5 ml. The measurements were carried out with a coaxial HPGe detector of 250 cm<sup>3</sup> volume, 64.8% relative efficiency (1332 keV). The source-detector distance was 42 mm from the detector surface. The obtained  $R_{\gamma_n+\gamma_m}$ -values for different solutions and compounds of Eu(III) are presented in Tab.1:

Tab.1:  $R_{\gamma_n+\gamma_m}$ -values of <sup>154</sup>Eu for different solutions and compounds

Compounds	$R_{\gamma_1+\gamma_2}$	$\sigma$
1M HClO <sub>4</sub>	1.1945	0.0043
0.01M HClO <sub>4</sub>	1.1946	0.0043
Eu(III)-DTPA	1.1519	0.0040
Eu(OH) <sub>3</sub> (c)	1.1176	0.0041
EuF <sub>3</sub> (c)	1.0964	0.0040

$R_{\gamma_n+\gamma_m}$ -values, reflecting the different physical-chemical status of Eu(III), are maximum for Eu<sup>3+</sup>-cationic forms and decrease due to complex or solid phase formations. It is thus shown that this approach allows to study the physical-chemical properties of <sup>154</sup>Eu.

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[1] D. V. Filosofov et. al, Appl. Rad. and Isot. **57**, 437 (2002). [2] F. D. Sowby, Radionuclides transformations. Energy and Intensity of Emissions. ICRP Publication 38. (1983).