Phantom measurements of ⁷⁴As and ¹⁸F at a Focus 120 small animal PET scanner

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Introduction: The positron emitter 74 As ($T_{1/2} = 17.8$ d) has been used in the 1950's as carrier-added arsenate for brain imaging with PET [1] and then disappeared with increasing development of isotope production and design of no-carrier-added (nca) radiopharmaceuticals. Now nca separation techniques for radioactive arsenic isotopes are available and labeling chemistry for interesting biomolecules that are suitable for molecular imaging like monoclonal antibodies (mab) is established [2]. Molecular imaging with sophisticated 74 As-labelled pharmaceuticals seems to be possible.

The aim of these phantom measurements (Fig.1) is to test the resolution of ⁷⁴As at the small animal PET camera Focus 120 and to compare the results with the resolution parameter of ¹⁸F. A plexiglas phantom was developed in the institute's workshop, Fig. 1.



Fig. 1:
Plexiglas phantom
with 3 drills of 2, 3
and 4 mm in
diameter, each.

Experimental: 300 mg of proton irradiated GeO₂ (3 μ A, 1 μ Ah) containing ⁷⁴As are dissolved in 2 ml of 5 M NaOH. The phantom is filled with a total volume of 1.5 ml of this solution corresponding to 750 kBq of ⁷⁴As (measured with HPGe detector, Fig. 2) and measured for 10 h (Fig. 3). The same phantom was filled with 10 MBq of ¹⁸F in water and measured for 1 h.

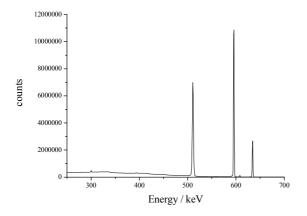


Fig.2: y-spectrum of ⁷⁴As with lines at 511, 595 and 635 keV

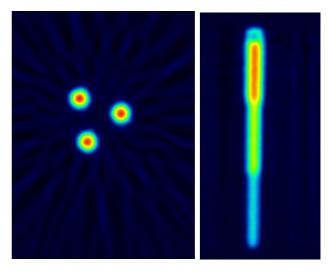


Fig.3: coronar and saggital PET images with 74 As in a phantom from Fig.1

Results: The comparative measurements of 74 As and 18 F show that 74 As is a useful isotope for PET imaging. Its resolution is close to that of 18 F (see Tab. 1) [3]. This is caused by the low mean β^+ -energy of only 128 keV of 74 As.

Tab. 1: Resolution of ⁷⁴As and ¹⁸F

Real size	⁷⁴ As FWHM	¹⁸ F FWHM
[mm]	[mm]	[mm]
4	3.80	3.71
3	3.11	3.02
2	2.76	2.62
$E_{\beta+}$ (mean) [kev]	128	242

References:

- Sweet, W.H. and G.L. Brownell, Localization of intracranial lesions by scanning with positron-emitting arsenic. J Am Med Assoc, 1955. 157(14): p. 1183-1188.
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- 3. Jennewein, M., et al., *A new method for radiochemical separation of arsenic from irradiated germanium oxide*. Appl Radiat Isot, 2005. **63**(3): p. 343-51.